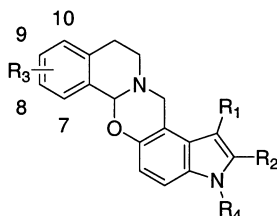


**Thomas M. Böhme,\* Corinne E. Augelli-Szafran, Hussein Hallak, Thomas Pugsley, Kevin Serpa, and Roy D. Schwarz:** Synthesis and Pharmacology of Benzoxazines as Highly Selective Antagonists at M<sub>4</sub> Muscarinic Receptors.

Page 3096. Entries in the first column of Table 1 are missing. The correct table is shown below.

**Table 1.** Binding Results of Benzoxazines



example	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>	R <sup>4</sup>	pK <sub>i</sub> <sup>a</sup>					formula <sup>c</sup>	mp (°C)
					M <sub>1</sub>	M <sub>2</sub>	M <sub>3</sub>	M <sub>4</sub>	M <sub>5</sub>		
<b>16</b>	CO <sub>2</sub> CH <sub>3</sub>	CH <sub>3</sub>	9-OMe	CH <sub>3</sub>	5.93	6.38	6.40	6.72	5.44	C <sub>23</sub> H <sub>24</sub> N <sub>2</sub> O <sub>4</sub>	194–195
<b>17</b>	CO <sub>2</sub> CH <sub>2</sub> CH=CH <sub>2</sub>	CH <sub>3</sub>	9-OMe	H	5.88	5.97	6.39	7.82	5.65	C <sub>24</sub> H <sub>24</sub> N <sub>2</sub> O <sub>4</sub> ·0.1H <sub>2</sub> O	188–190
<b>18</b>	CO <sub>2</sub> CH <sub>2</sub> CH=CH <sub>2</sub>	CH <sub>3</sub>	9-OMe	( <i>p</i> -OMe)Bz	6.70	6.62	6.17	7.14	6.30	C <sub>32</sub> H <sub>32</sub> N <sub>2</sub> O <sub>5</sub> ·0.04H <sub>2</sub> O	175–176
<b>19</b>	CO <sub>2</sub> CH <sub>2</sub> C <sub>6</sub> H <sup>15</sup>	CH <sub>3</sub>	9-OMe	H	5.93	5.87	5.86	7.14	5.95	C <sub>28</sub> H <sub>26</sub> N <sub>2</sub> O <sub>4</sub>	176–178
<b>20</b>	CO <sub>2</sub> (CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	CH <sub>3</sub>	9-OMe	H	5.88	5.63	5.99	7.17	5.60	C <sub>25</sub> H <sub>28</sub> N <sub>2</sub> O <sub>4</sub>	177–178
<b>21</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	CH <sub>2</sub> CH <sub>3</sub>	9-OMe	H	5.66	5.87	7.04	8.40	5.51	C <sub>24</sub> H <sub>26</sub> N <sub>2</sub> O <sub>4</sub>	183–184
<b>22</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	CH <sub>2</sub> CH <sub>3</sub>	9-OMe	benzyl	6.44	6.43	7.30	7.01	6.32	C <sub>31</sub> H <sub>32</sub> N <sub>2</sub> O <sub>4</sub> ·0.04H <sub>2</sub> O	75–85
<b>23</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	CH <sub>2</sub> CH <sub>3</sub>	9-OMe	CH <sub>3</sub>	5.82	6.15	6.66	8.52	5.51	C <sub>25</sub> H <sub>28</sub> N <sub>2</sub> O <sub>4</sub> ·0.06H <sub>2</sub> O	172–174
<b>24</b>	<b>CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub></b>	<b>(CH<sub>2</sub>)<sub>2</sub>CH<sub>3</sub></b>	<b>9-OMe</b>	<b>H</b>	<b>5.69</b>	<b>5.97</b>	<b>7.15</b>	<b>8.70</b>	<b>5.65</b>	<b>C<sub>25</sub>H<sub>28</sub>N<sub>2</sub>O<sub>4</sub>·0.11H<sub>2</sub>O</b>	<b>194–196</b>
<b>25</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	(CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	9-SMe	H	6.06	5.84	7.00	9.00	5.62	C <sub>25</sub> H <sub>28</sub> N <sub>2</sub> O <sub>3</sub> S·0.02H <sub>2</sub> O	185–186
<b>26</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	(CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	9-SMe	CH <sub>3</sub>	5.48	5.42	6.64	8.70	5.77	C <sub>26</sub> H <sub>30</sub> N <sub>2</sub> O <sub>3</sub> S·0.27H <sub>2</sub> O	139–143
<b>27</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	(CH <sub>2</sub> ) <sub>2</sub> CH <sub>3</sub>	9-OMe	CH <sub>3</sub>	5.88	6.29	7.15	8.70	5.47	C <sub>26</sub> H <sub>30</sub> N <sub>2</sub> O <sub>4</sub>	150–151
<b>28</b>	<b>CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub></b>	<b>(CH<sub>2</sub>)<sub>3</sub>CH<sub>3</sub></b>	<b>9-OMe</b>	<b>H</b>	<b>4.88</b>	<b>6.47</b>	<b>6.82</b>	<b>9.00</b>	<b>4.95</b>	<b>C<sub>26</sub>H<sub>30</sub>N<sub>2</sub>O<sub>4</sub></b>	<b>180–182</b>
<b>29</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	(CH <sub>2</sub> ) <sub>3</sub> CH <sub>3</sub>	9-SMe	H	6.06	5.66	6.47	8.52	5.83	C <sub>26</sub> H <sub>30</sub> N <sub>2</sub> O <sub>3</sub> S	159–161
<b>30</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	(CH <sub>2</sub> ) <sub>4</sub> CH <sub>3</sub>	9-OMe	H	4.69	5.14	6.34	7.96	5.27	C <sub>27</sub> H <sub>32</sub> N <sub>2</sub> O <sub>4</sub>	170–172
<b>31</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	(CH <sub>2</sub> ) <sub>5</sub> CH <sub>3</sub>	9-OMe	H	5.51	5.34	6.55	7.85	>4.55	C <sub>28</sub> H <sub>34</sub> N <sub>2</sub> O <sub>4</sub>	157–160
<b>32</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	CH <sub>3</sub>	9-C1	H	5.29	5.49	5.56	6.44	5.27	C <sub>22</sub> H <sub>21</sub> ClN <sub>2</sub> O <sub>3</sub> ·0.7H <sub>2</sub> O	196–197
<b>33</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	CH <sub>3</sub>	7-C1	H	>5.36	5.61	5.29	5.71	5.32	C <sub>22</sub> H <sub>21</sub> ClN <sub>2</sub> O <sub>3</sub> <sup>d</sup>	188–189
<b>34</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	CH <sub>3</sub>	9-SMe	H	5.75	5.66	6.64	8.15	5.71	C <sub>25</sub> H <sub>24</sub> N <sub>2</sub> O <sub>3</sub> S·0.02H <sub>2</sub> O	188–189
<b>35</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	CH <sub>3</sub>	9-OH	H	>4.66	>4.57	4.59	>4.91	>4.55	C <sub>22</sub> H <sub>22</sub> N <sub>2</sub> O <sub>4</sub> <sup>e</sup>	241–243
<b>36</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	CH <sub>3</sub>	9-OMe	CH <sub>2</sub> CH <sub>3</sub>	5.88	6.66	6.58	6.96	5.60	C <sub>25</sub> H <sub>28</sub> N <sub>2</sub> O <sub>4</sub> <sup>f</sup>	173–175
<b>37</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	CH <sub>3</sub>	9-OMe	CH <sub>3</sub>	5.70	6.29	6.04	7.82	5.43	C <sub>24</sub> H <sub>26</sub> N <sub>2</sub> O <sub>4</sub> ·0.1H <sub>2</sub> O	174–176
<b>38</b>	CONCH <sub>3</sub> C <sub>2</sub> H <sub>5</sub>	CH <sub>3</sub>	9-OMe	H	4.66	5.39	4.81	5.01	4.71	C <sub>24</sub> H <sub>27</sub> N <sub>3</sub> O <sub>3</sub> <sup>g</sup>	162–164
<b>39</b>	CH <sub>2</sub> C <sub>5</sub> H <sub>6</sub>	CH <sub>3</sub>	9-OMe	H	5.60	5.27	5.69	5.87	5.95	C <sub>27</sub> H <sub>26</sub> N <sub>2</sub> O <sub>2</sub> ·0.04H <sub>2</sub> O	196–198
<b>40</b>	oxadiazole <sup>b</sup>	CH <sub>3</sub>	9-OMe	H	5.12	5.66	5.51	7.21	5.21	C <sub>23</sub> H <sub>22</sub> N <sub>4</sub> O <sub>3</sub> ·0.19H <sub>2</sub> O	195–197
<b>41 (PD102807)</b>	CO <sub>2</sub> CH <sub>2</sub> CH <sub>3</sub>	CH <sub>3</sub>	9-OMe	H	5.88	6.27	6.59	8.15	5.47	C <sub>28</sub> H <sub>24</sub> N <sub>2</sub> O <sub>4</sub>	190–192

<sup>a</sup> The affinity of these compounds for the five human receptor subtypes (M<sub>1</sub>–M<sub>5</sub>) was determined by [<sup>3</sup>H]-NMS binding using membranes from transfected Chinese hamster ovarian (CHO) cells. All compounds were tested at least three times, and the value shown is given as the geometric mean. The standard error of the mean (SEM) for all compounds tested ranged from 6% to 18%. The complete protocol is described by Dörje et al.<sup>12</sup> and Buckley et al.<sup>13</sup> The K<sub>i</sub> values of the test compounds were derived from IC<sub>50</sub> values using the Cheng–Prusoff equation,  $K_i = IC_{50}/(1 + L/K_d)$ , with the radioligand concentration  $L = 0.1$  nM and the following equilibrium dissociation constants  $K_D$  of [<sup>3</sup>H] NMS, determined in previous saturation binding experiments (nM): M<sub>1</sub> (1.28); M<sub>2</sub> (0.85); M<sub>3</sub> (0.94); M<sub>4</sub> (3.10); M<sub>5</sub> (0.79).<sup>30</sup>

<sup>b</sup> Oxadiazole = 3-methyl[1,2,4]oxadiazol-5-yl. <sup>c</sup> All compounds have analytical results within ±0.4% of theoretical values unless otherwise noted. Some difficulty was found in obtaining combustion analysis in the indicated compounds because of the propensity of these compounds to retain solvents. <sup>d</sup> High mass: calcd 397.1319; found 397.1325. <sup>e</sup> High mass: calcd 379.1658; found 379.1651. <sup>f</sup> High mass: calcd 421.2127; found 421.2122. <sup>g</sup> High mass: calcd 406.2130; found 406.2137.

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